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the Austrian foreign minister who was the leader of the Congress of Vienna; claimed that he was guided by the principal of legitimacy Conservatism based on tradition and social stability/ favored obedience to political authority and believed that organized religion was crucial to order in society

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Part 3: Aqueous Ammonia Solution. Instructor Prep: At the beginning of lab prepare a stock solution of aqueous ammonia. Add 4 drops of concentrated 15 M  $\text{NH}_3$  (aq) and 3 drops of phenolphthalein to a 150-mL (medium) beaker, top it up with 100-mL of distilled water, and mix with a stirring rod. Label the beaker and place it on the front desk.

### 12: Equilibrium and Le Chatelier's Principle (Experiment ...

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7. In both of these reactions, a precipitate was formed. A precipitate is a solid that forms in a reaction but does not dissolve. The precipitate will typically sink to the bottom of the test tube. Write the chemical formula and name for the precipitate you found in this lab. 8. Mixing two ionic compounds does not always produce a reaction.

### Double Replacement Reaction Name

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2 to sugar Answer the following questions as you read modules 7.10–7.11: 1. The Calvin cycle is cyclic because a. CO<sub>2</sub> is constantly acquired during carbon fixation. b. it constantly makes G3P. c. RuBP is regenerated in the final chemical reaction. d. NADPH and ATP are used from the light reactions. 2.

### **Chapter 7: Photosynthesis: Using Light to Make Food**

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