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bonds that hold atoms together beha....
When two hydrogen atoms bond, the
posit.... An ionic bond is a bond that
forms betw.... In which type of bond do
atoms share el.... flexible springs.
negative electron of the other atoms.

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5 terms. quizletteloganTEACHER. DRHS: Physical Science: Chapter 6 Formulas for Ionic Compounds. Formula Unit. Oxidation Number. Determining the Formula for an Ionic Co.... Polyatomic Ions. Simplest ratio of the ions

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represented in an ionic compound... The charge of monoatomic (1 atom) ion (same thing as the charg....

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Study the electron dot diagrams for lithium, carbon, flourine, and neon in figure 6-1. Choose the statement that

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correctly identifies the most stable of the elements. A) lithium is the most stable element because it has to lose only one electron to achieve a stable configuration. B) carbon is the most stable element because it can form four bonds.

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conductor. electrical field. direct...
process of transferring charge between
objects by touch.... process of
rearranging electrons on a neutral object
by bringi....

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Section 6.4 The Structure of Metals
(pages 176–181) This section discusses
metallic bonds and the properties of
metals. It also explains how the
properties of an alloy are controlled. <http://mrmicco.weebly.com/uploads/1/2/5/8/12584959/psgs1064b.pdf>

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